

**MEMS 1051 Applied Thermodynamics  
Spring 2018**

**Course Coordinator:** Dr. Sam Ghalambor  
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**Office Hours:** Mondays 9:00(AM)-10:00(AM), or by appointment

**Credits and contact hours (lecture/lab):** 3 Credits, 3 Contact Hours (lecture)

**Designated as 'Required' or 'Elective' Course:** Required

**Course description:** Thermodynamic processes involving energy and entropy changes in real and ideal gases, vapors, and liquids, and mixtures of those fluids. Basic thermodynamic cycles (vapor and gas power, refrigeration, and heat pumps). Air-water mixtures and psychrometrics. Thermodynamics of reactions and combustion. Applications to refrigeration, power plants, and internal combustion engines.

**Prerequisite and Co-requisite:** MEMS 0051

**Textbook:**

- 1) "Fundamentals of Thermodynamics," Claus Borgnakke and Richard E. Sonntag, 8th Ed., John Wiley and Sons.
- 2) "Thermodynamics, an engineering approach", Yunus A. Cengel and Michael A. Boles 5<sup>th</sup> Ed., McGraw-Hill Education

**Reference:** "Fundamentals of Engineering Thermodynamics 5<sup>th</sup> ed.", Michael J. Moran, Howard N. Shapiro

**Other optional materials:** Engineering Equation Solver (EES) software, v.10

**Course Objectives:**

1. To develop additional thermodynamic tools for analyzing processes and systems.
2. To understand entropy change and entropy generation of processes and systems.
3. To learn how to apply principles of thermodynamics to analyze practical energy systems.
4. To learn basic thermodynamic cycles and their application to real systems
5. To become familiar with the practice of psychrometrics.
6. To learn how to analyze the thermodynamics of reactions and combustion processes.
7. To learn to evaluate energy systems for efficient energy utilization, and to understand the impact of energy utilization on materials and the environment.

**Course learning outcomes/expected performance criteria:**

1. Conservation of Energy for Open Systems (70%)

2. Conservation of Energy for Closed Systems (70%)
3. Calculation and Use of Entropy (70%)
4. First and Second Law Efficiencies (70%)
5. Power Cycle Analysis (70%)
6. Heat Pump/Refrigeration Cycle Analysis (70%)
7. Calculation of Properties for Pure Substances and Simple Mixtures (70%)
8. Combustion and Reactions (70%)

**Course topics and lecture hours devoted to each topic:**

1. First law and thermodynamic properties
2. Conservation of energy for a control volume
3. The second law and entropy
4. Phase-change power cycles
5. Gas power cycles
6. Refrigeration and heat pump cycles
7. Project working classes
8. Nonreacting ideal gas mixtures and psychrometrics
9. Thermodynamics of combustion and reactions
10. Exams and Review

**Grading Policy**

The course grade will be determined based on the following contributions:

Midterm I	25%
Midterm II	25%
Final Exam	40%
Homework	10%

**Disability Services**

If you have a disability for which you are or may be requesting an accommodation, you are encouraged to contact me or admin Staff for accommodation.

**Academic Integrity**

All students are expected to adhere to the standards of academic honesty. Any student engaged in cheating, plagiarism, or other acts of academic dishonesty would be subject to disciplinary action. Any student suspected of violating this obligation for any reason during the semester will be required to participate in the procedural process, initiated at the instructor level, as outlined in the University Guidelines on Academic Integrity. This may include, but is not limited to the confiscation of the examination of any individual suspected of violating the University Policy.